

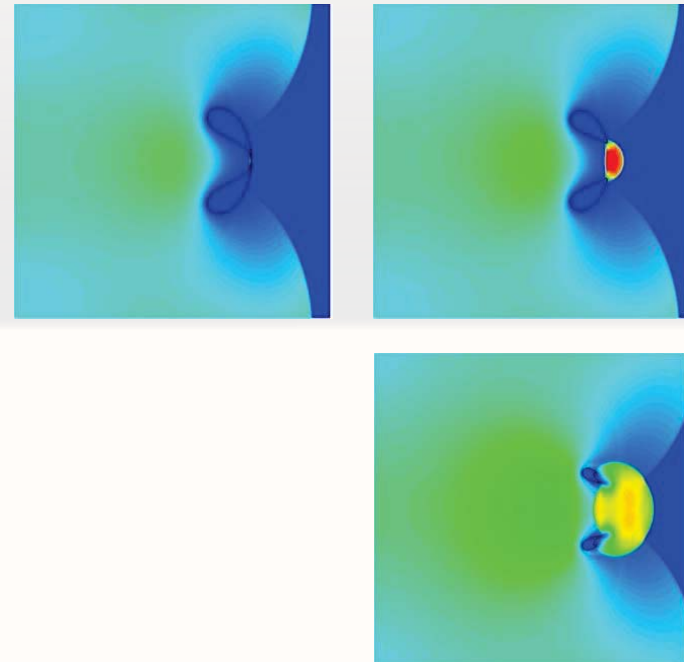
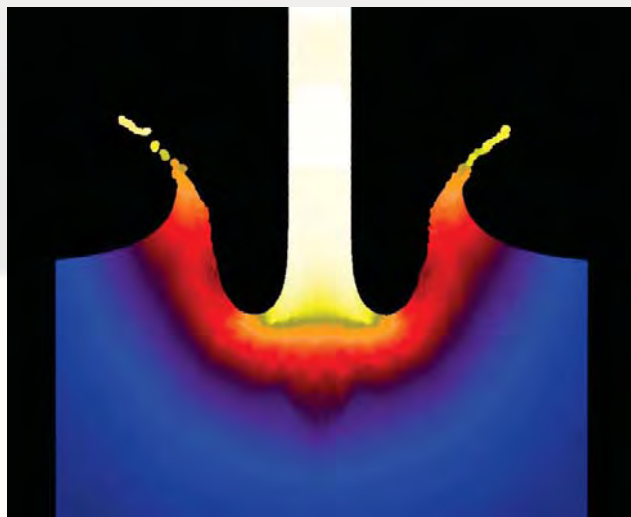
# **Explosion dynamics of propellants and laser ablation of metals**

**Workshop on Energetics-Past and Present  
Hong Kong Science and Technology Parks  
December 8, 2010**

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Head of Extreme Energy Laboratory  
Seoul National University**

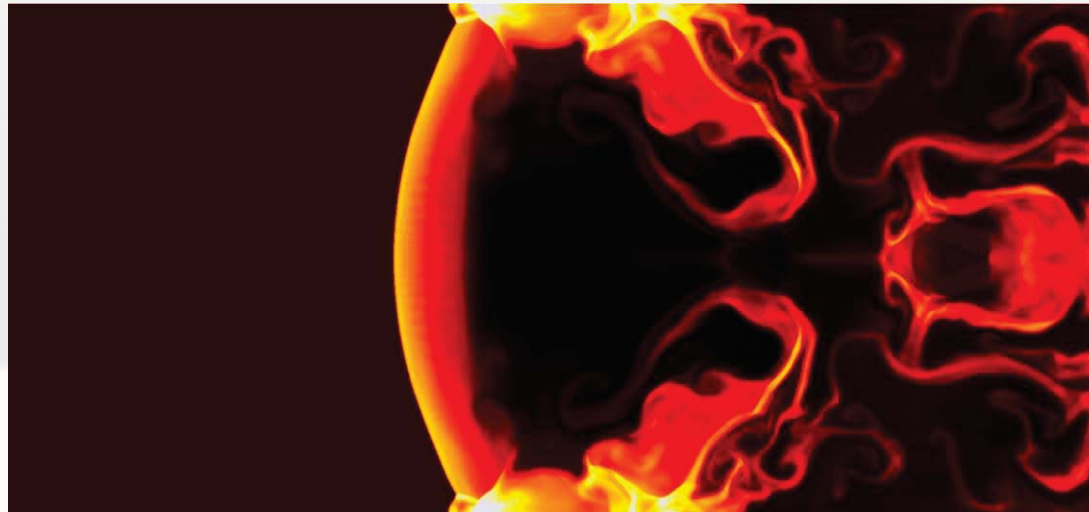
- **Hydrodynamic Shock Physics code** for energetic materials
- **Laser** induced detonation of metals
- Phenomena of great interest include
  - **Deflagration and detonation** of high energy materials
  - High energy thermo-fluid simulation
  - Explosion, Impact, Jet formation



High strain rate phenomena of multi-material interactions

# Cookoff and Flame acceleration

- Rocket propellant (Liquid and Solid) flame acceleration due to external stimuli (shock wave and thermal load - **cookoff**)
  - Deflagration and Detonation
  - Accidents and failures
  - Preventions and insensitive energetic materials



Shock-induced Detonation Transition (Ethylene-Air)

# Thermal decomposition of Propellant

- **AP/HTPB Reaction Model**

- **Modified 2-step Beckstead model**

- **AP 88%, HTPB 12%**
    - **Mixed combustion (premixed flame, primary diffusion flame, final diffusion flame)**

$R_1$  : AP (X) → decomposition product (Z)

$R_2$  :  $\beta Z$  + binder (Y) → final product

( $\beta = 88/12 = 7.33$ )

## 2-Step Kinetics

$$\frac{\partial \rho Y_X}{\partial t} = r_X = -R_1 \quad \frac{\partial \rho Y_Y}{\partial t} = r_Y = -R_2$$

$$\frac{\partial \rho Y_Z}{\partial t} = r_Z = R_1 - 7.51R_2$$

$$\rho Y_{product} = \rho - (\rho Y_X + \rho Y_Y + \rho Y_Z)$$

## Species evolution

$$R_1 = \rho Y_X Z_1 P_0^{1.744} \exp(-E_{a,1} / RT)$$

$$R_2 = \rho Y_Y \rho Y_Z Z_2 P_0^{1.75} \exp(-E_{a,2} / RT)$$

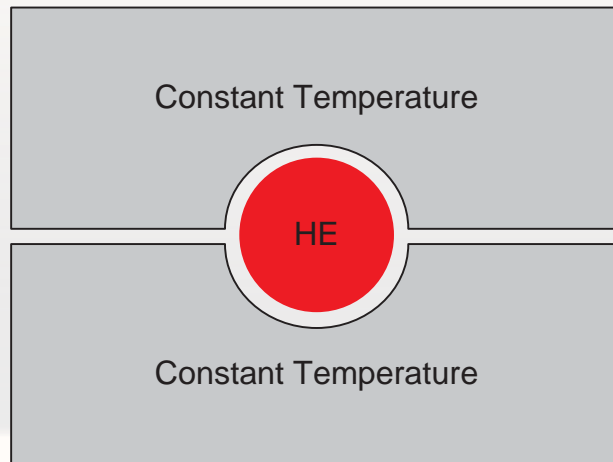
## Reaction Rate

Reaction Step	Z	E(KJ/mol)	Reaction Order	Heat of Reaction Q (kJ/kg)
1	800	137.18	1	-297
2	1100	178.75	2	+9643.2

## Reaction coefficients

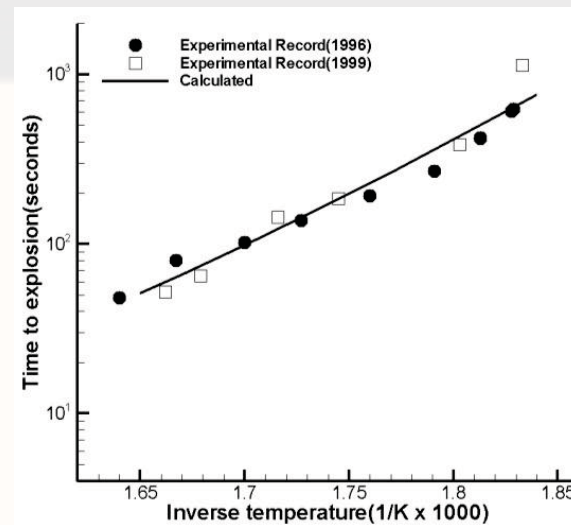
# Thermal decomposition

- One dimensional time to explosion(ODTX)
  - Kinetics verification apparatus
  - Constant volume explosion analysis of sample of 1.27cm in diameter: time to explosion is measured (LLNL)



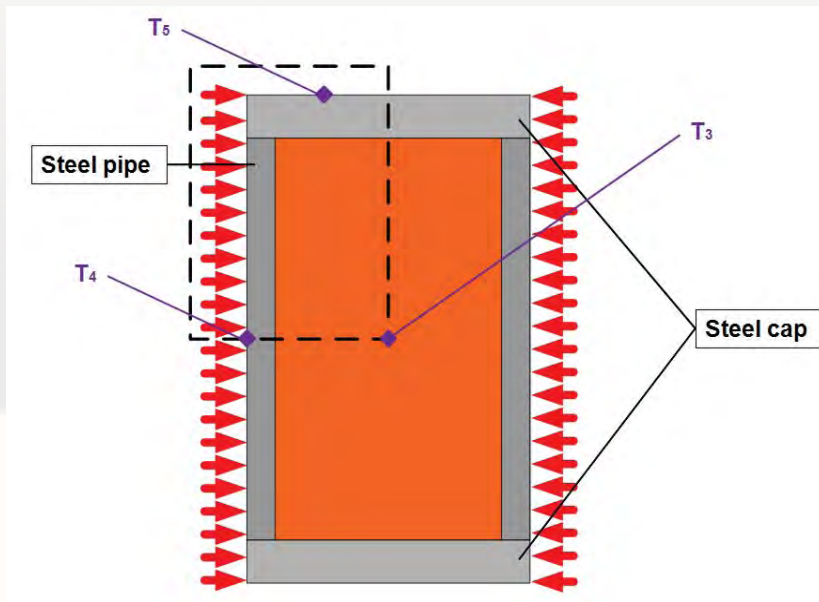
$$\rho C \frac{\partial T}{\partial t} = -\nabla \cdot \vec{q} - \sum_{j=1}^N R_j q_j$$

$$\rho \frac{\partial Y_i}{\partial t} = r_i$$

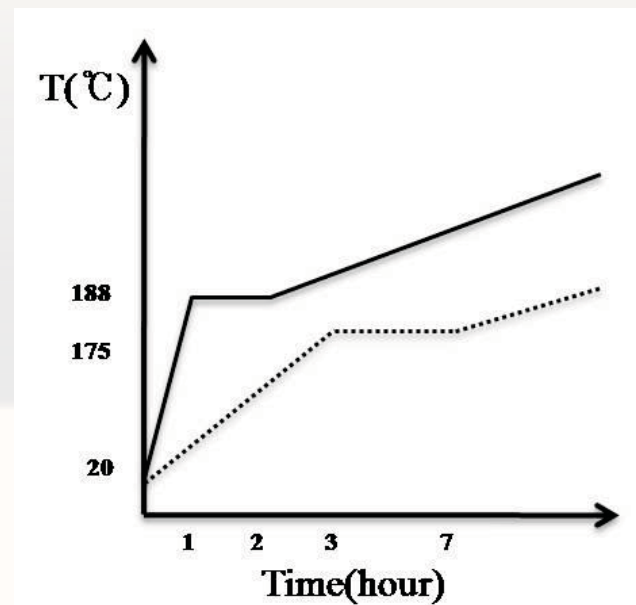


# AP/HTPB slow cookoff experiment

- **2 AP/HTPB Propellant (AP 88%, HTPB 12%) Tests**
  1. 188 °C/hour heat, 1 hour soak, and 6 °C/hour heat (2006)
  2. 175 °C/3hours, 4 hour soak, 1.8 °C/hour (2009)



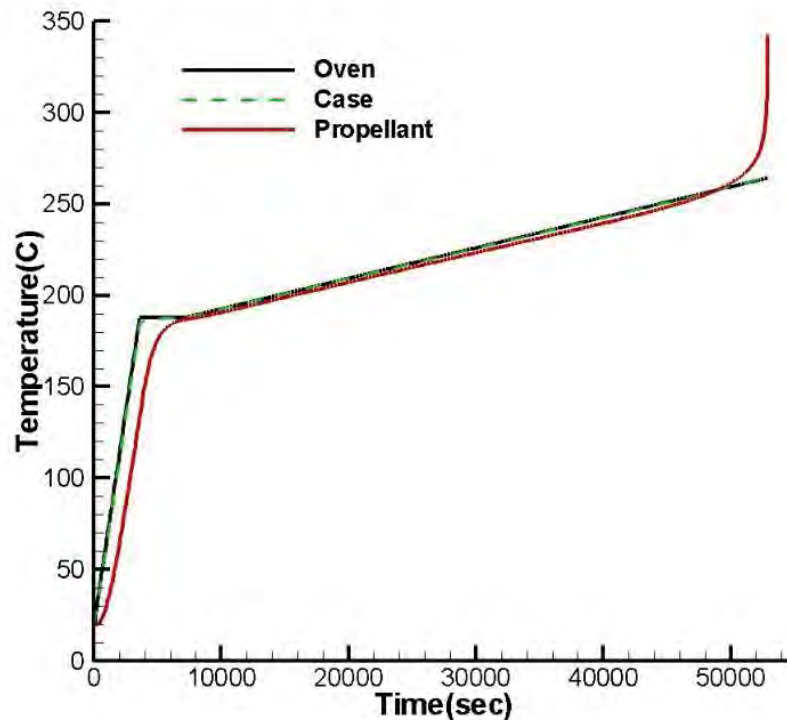
Modeling schematics



Control TC comparisons of two experiments

# AP/HTPB cookoff results

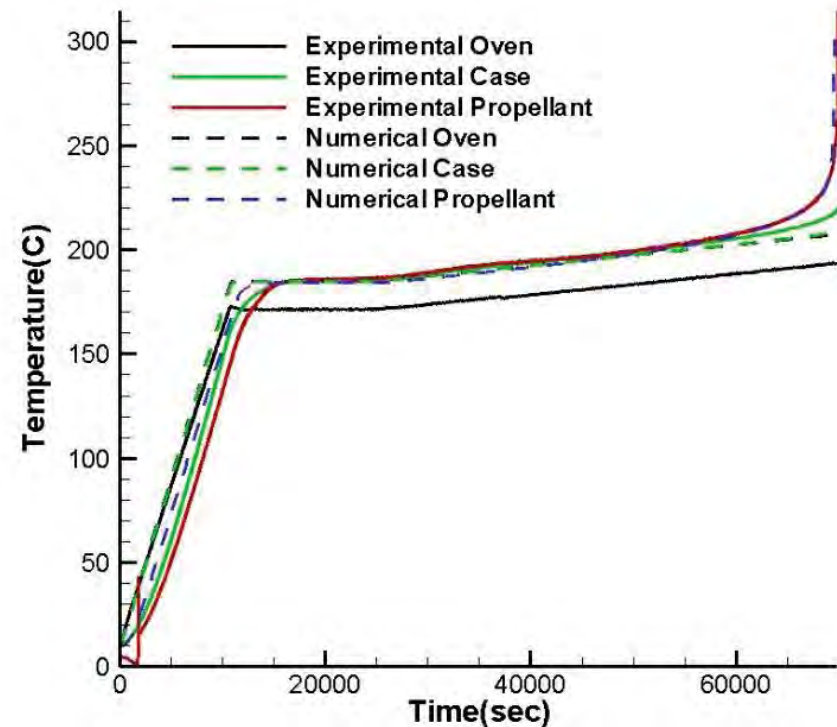
- Time and Temperature of explosion predictions



**TEST 1 (2006)**

Experiment ~ 15 hours

Calculation ~ 14.7 hours



**TEST 2 (2009)**

Experiment ~ 19.4 hours

Calculation ~ 19.4 hours

# AP/HTPE Insensitive Propellant

- **AP-AN/BuNENA-HTPE**
  - **AP/HTPB's lower decomposition temperature**
    - AP/HTPB decomposes at 250°C while AP/HTPE decomposes at 150°C
    - AP/HTPE meets safety requirement
  - **AP/HTPE's thermal properties**
    - Modified AP/HTPB meets preliminary AP/HTPE model



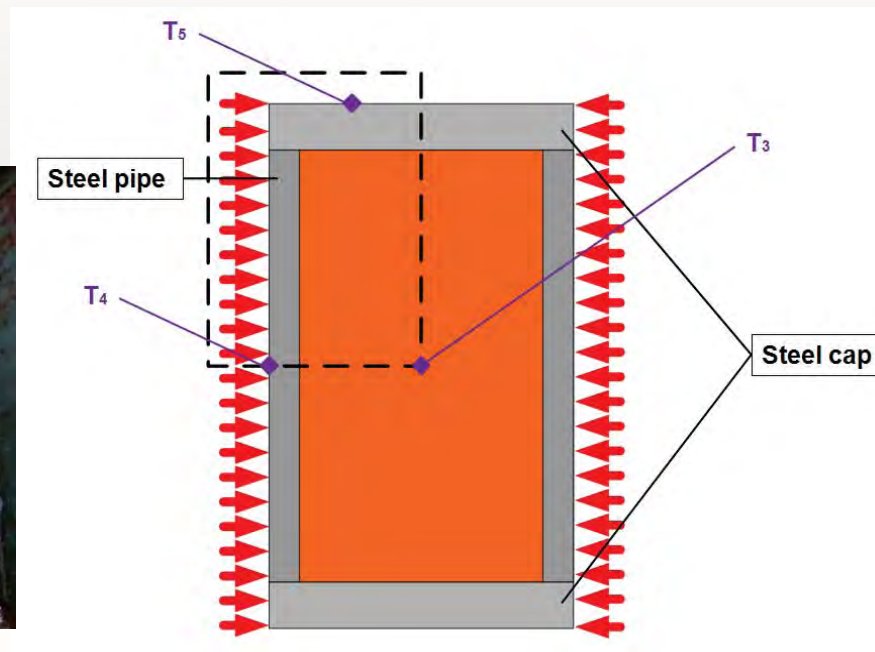


# AP/HTPE slow cookoff test

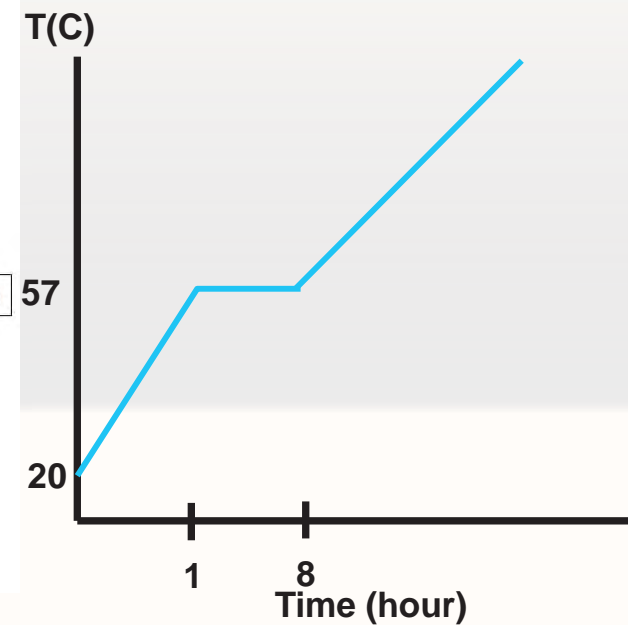
- **AP/HTPE Propellant**
  - AP 88%, HTPE 12%
  - Heat 57 °C/3 hours, soak for 7 hours, 3.3 °C/1 hour heat



Experiment (2009)



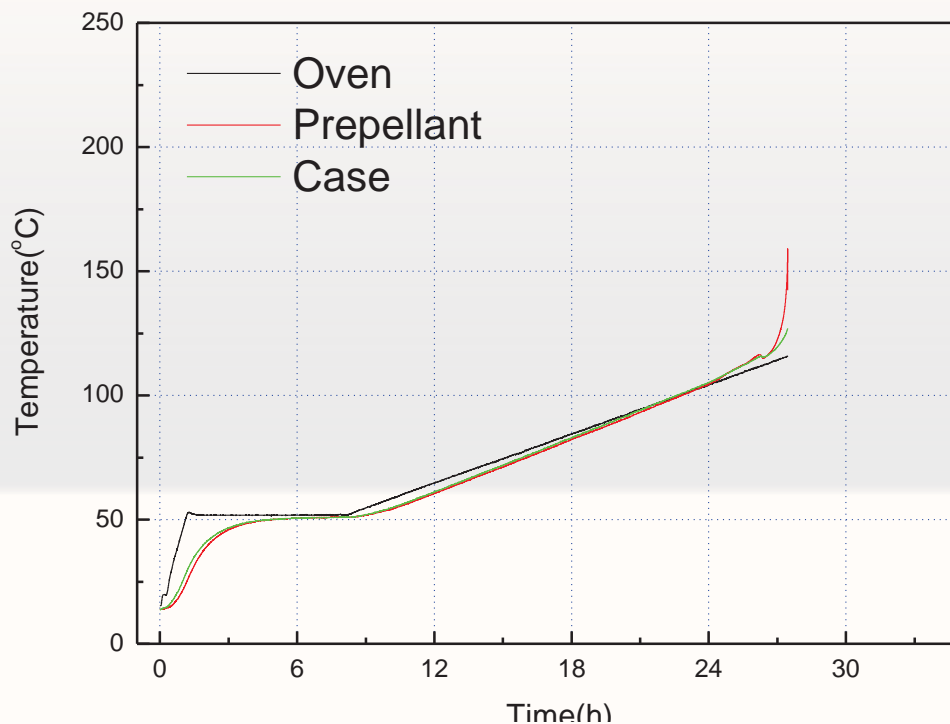
Simulation setup



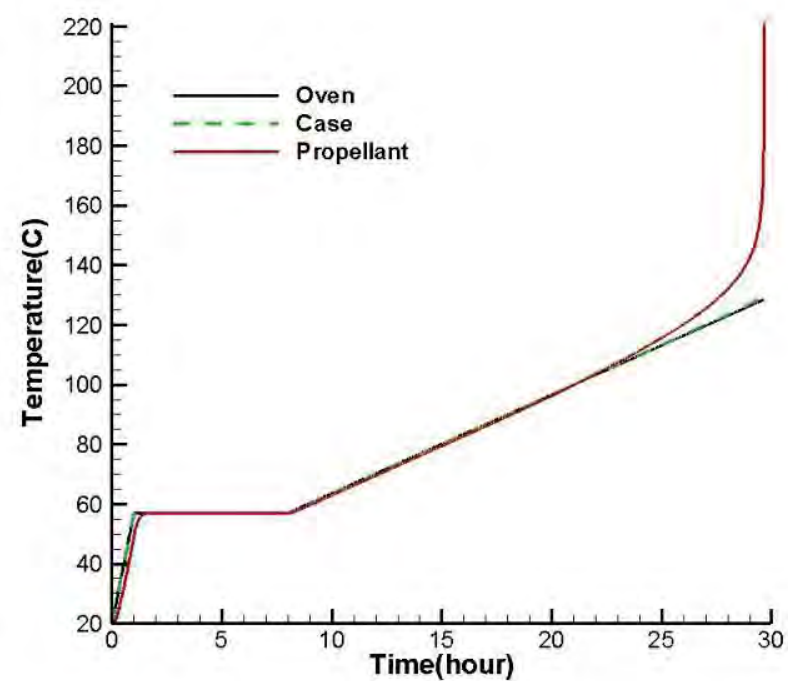
Control TC

# AP/HTPE slow cookoff calculated

- **Thermal explosion time-temperature comparison**
  - Measured : ~ 27.5 hours
  - Calculated : ~ 28.5 hours



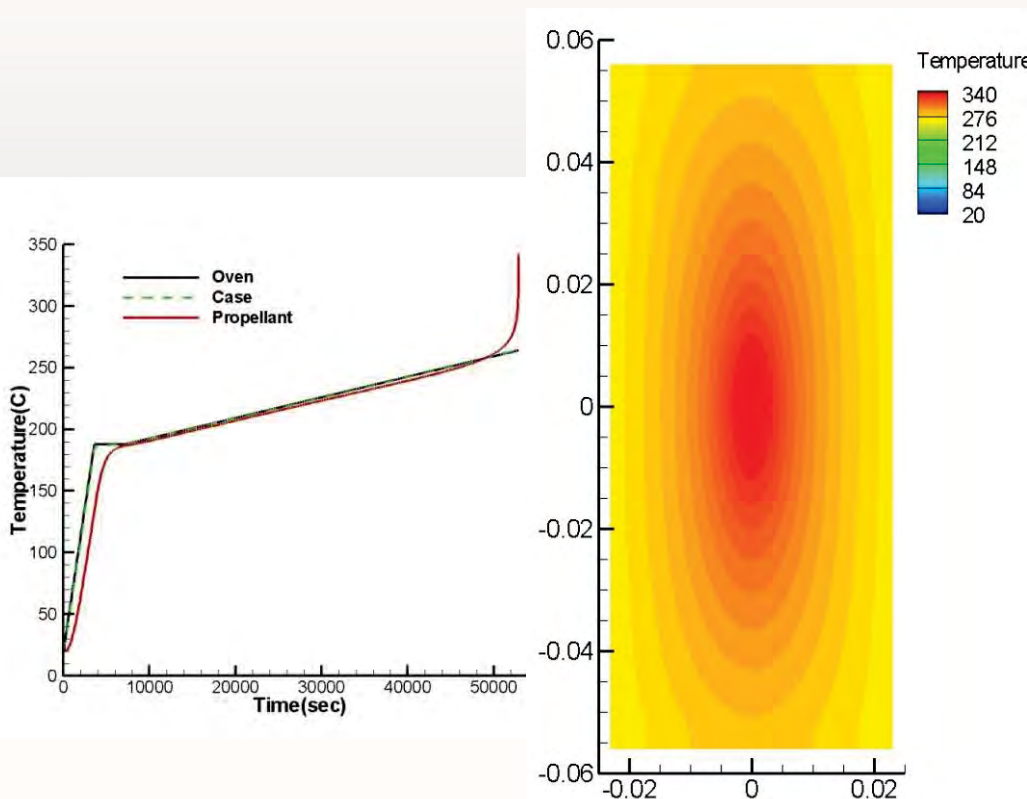
**Experiment (2009)**



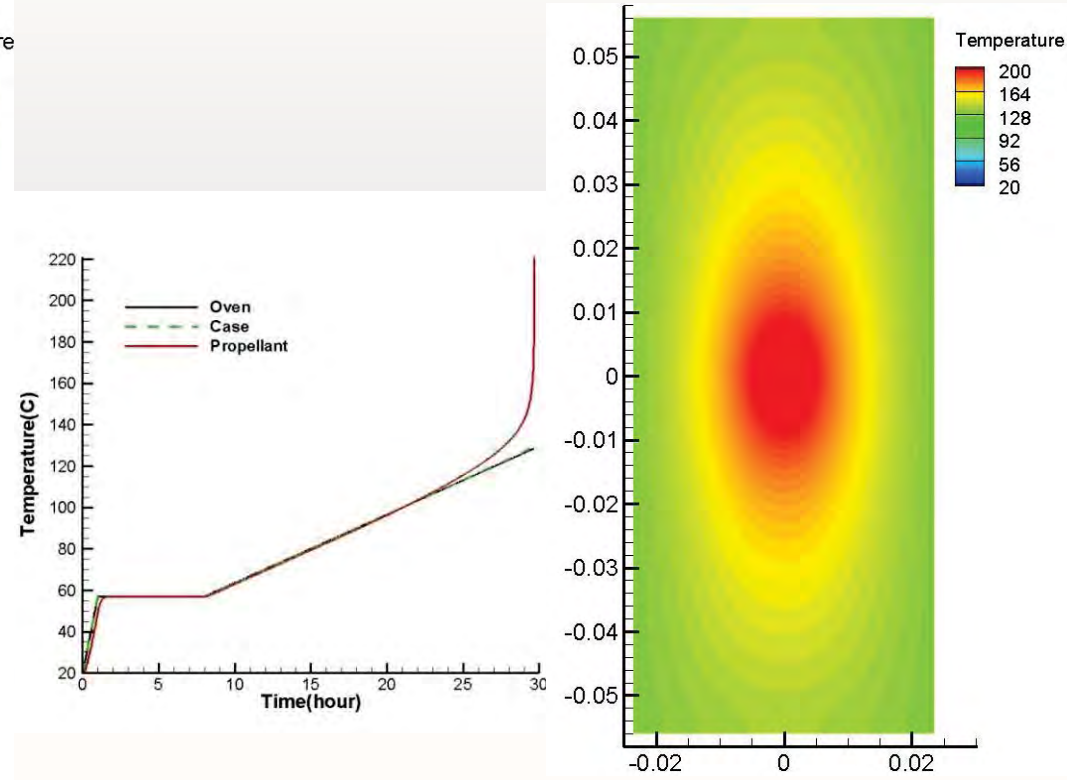
**Calculated**

# Thermal decomposition of AP propellants

- AP/HTPB, AP/HTPE
  - Runaway for AP/HTPB is at 250
  - Runaway for AP/HTPE is at 150°C
  -



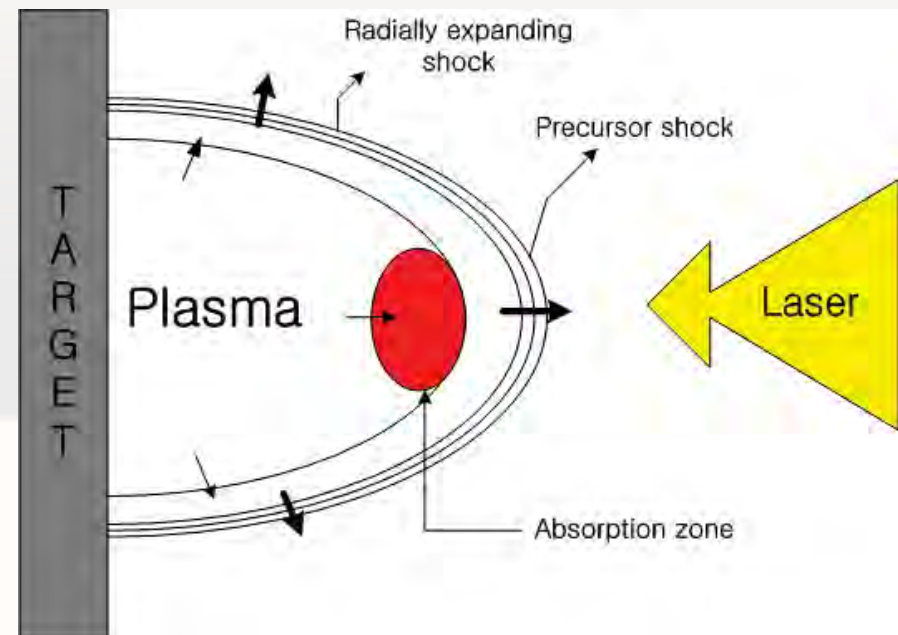
AP/HTPB



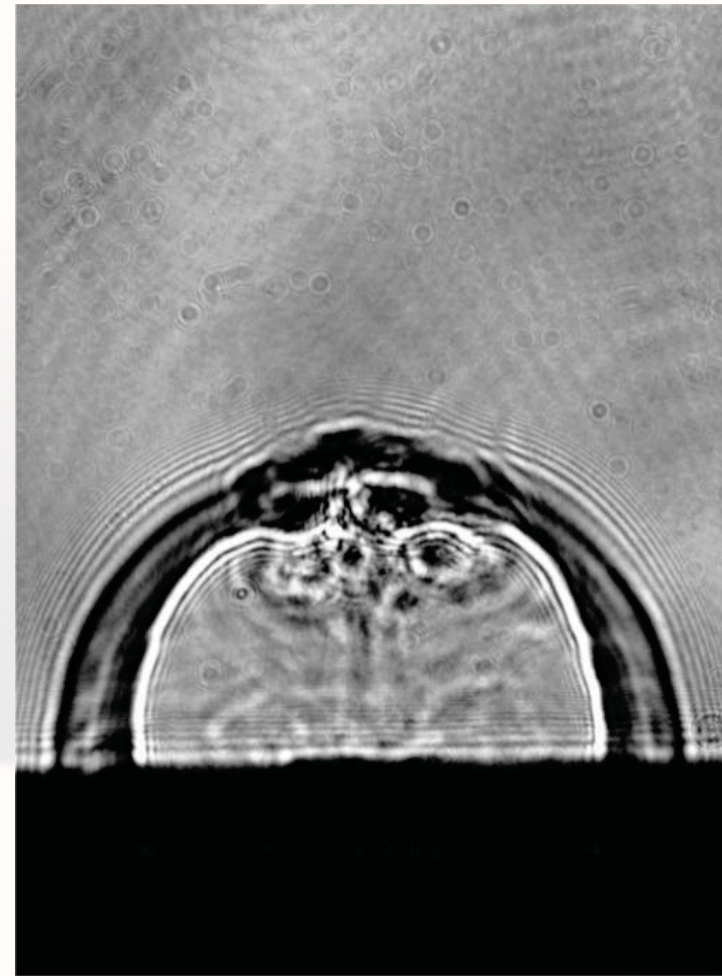
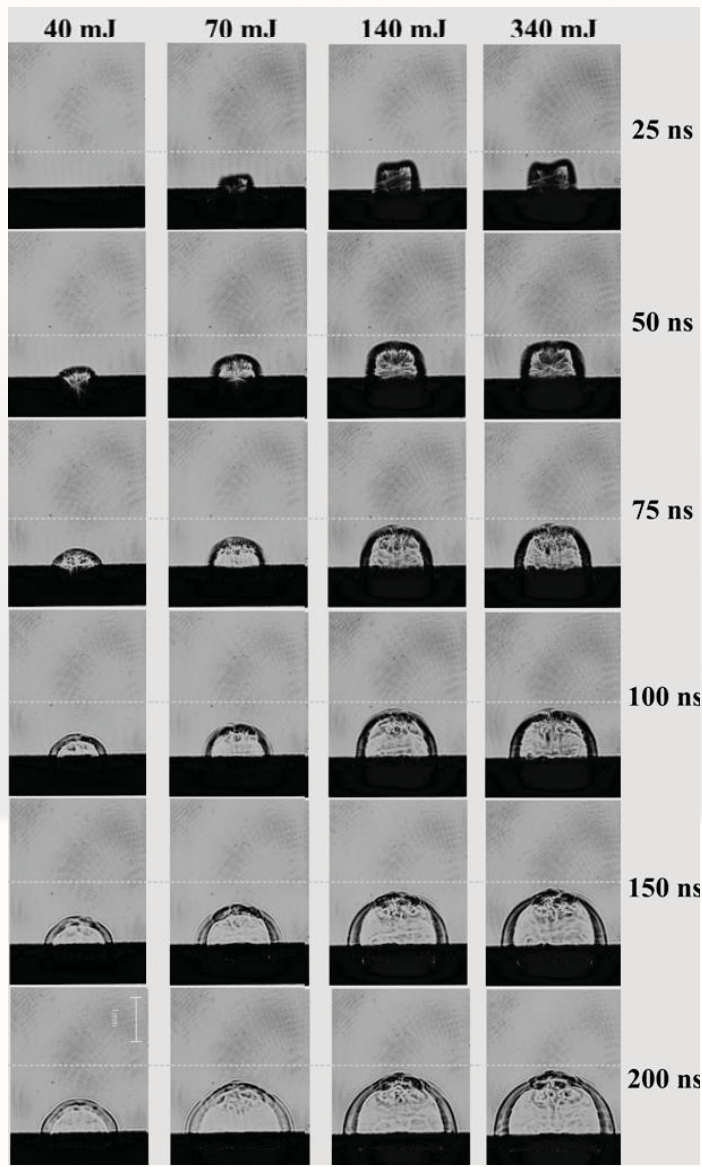
AP/HTPE

# Laser supported detonation

- **Laser ablation**
  - Laser induced surface reaction by the beamed energy
  - Plasma generation followed by shock wave propagation
- **Applications**
  - Chemical analysis
  - Material Processing
  - Laser propulsion, etc



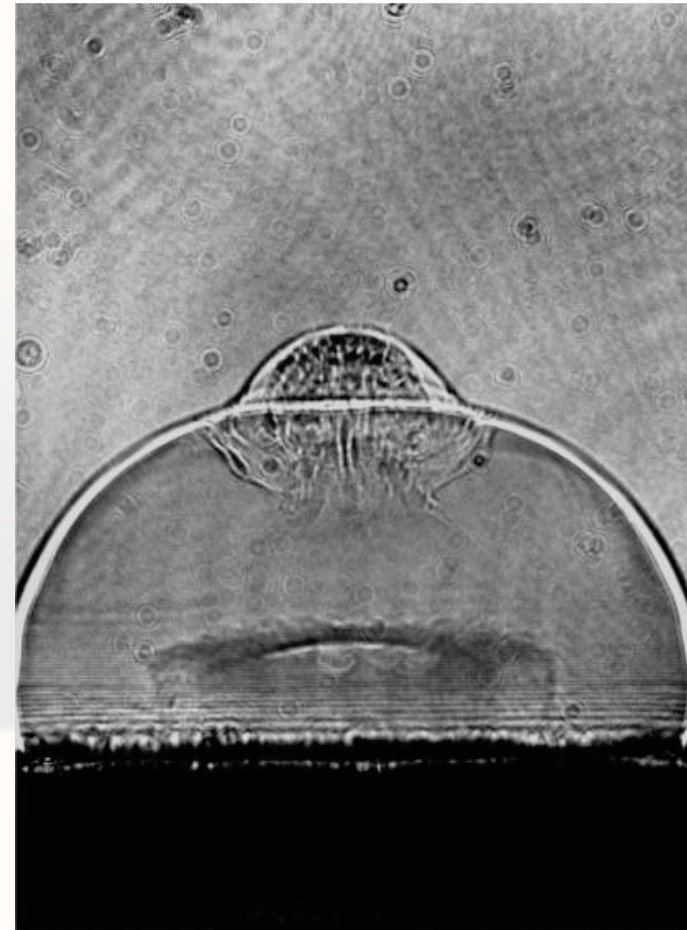
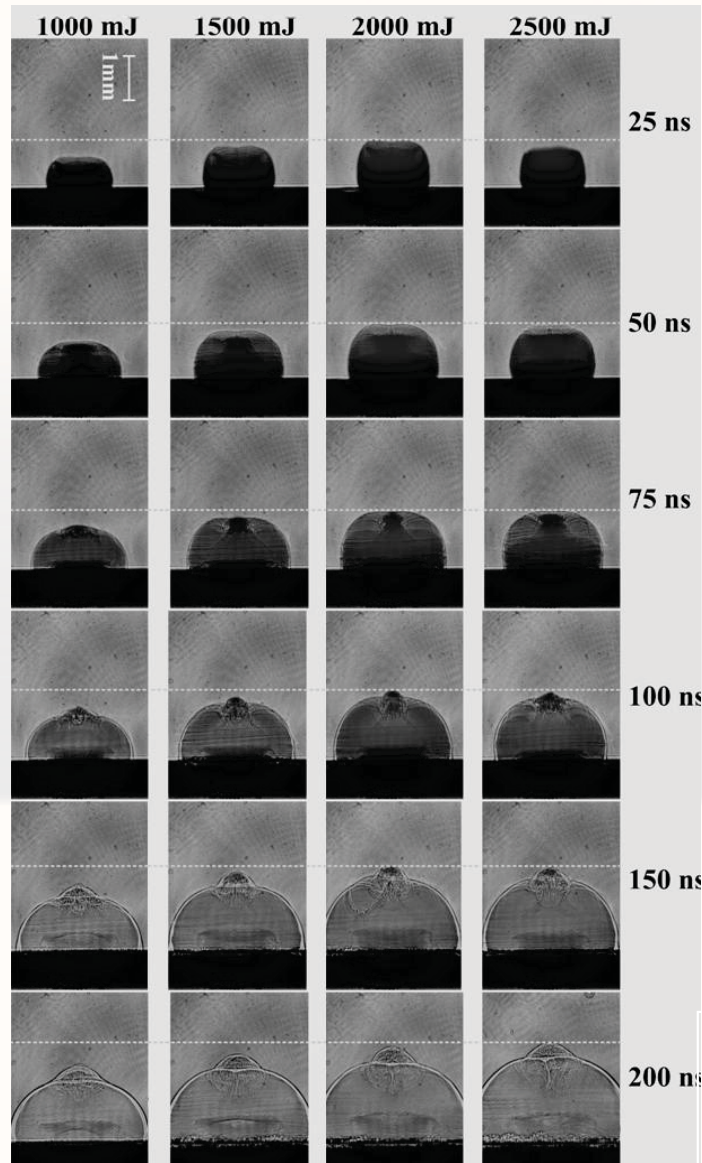
# Shadowgraph (40–340 mJ) 1 atm



- Ionized and shock fronts coupled without chemical reaction

<Increasing delay times in 1 atm over an Al target>

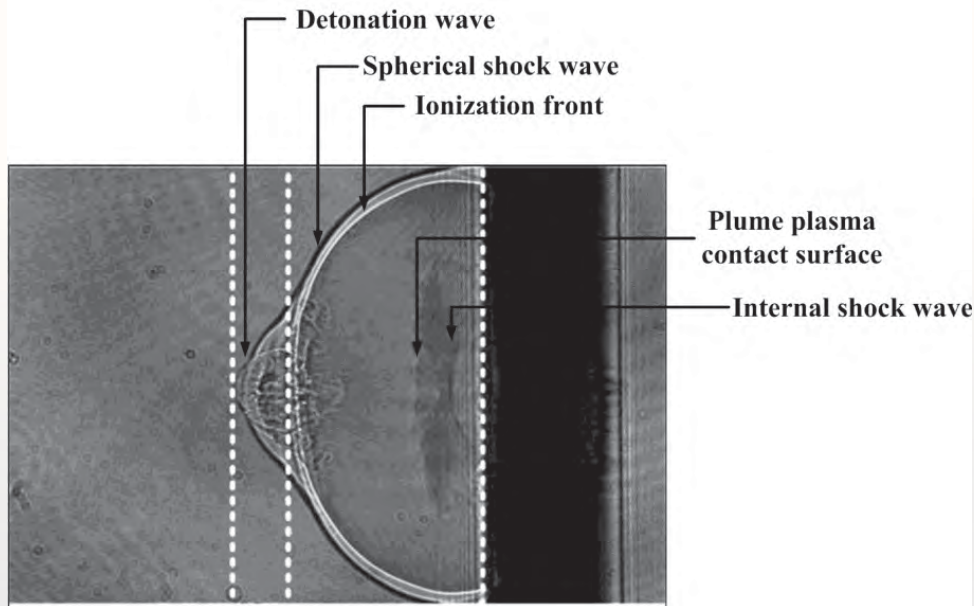
# Shadowgraph (1000–2500 mJ)



- Separation of shock and detonation fronts

<Increasing delay times in 1 atm over an Al target>

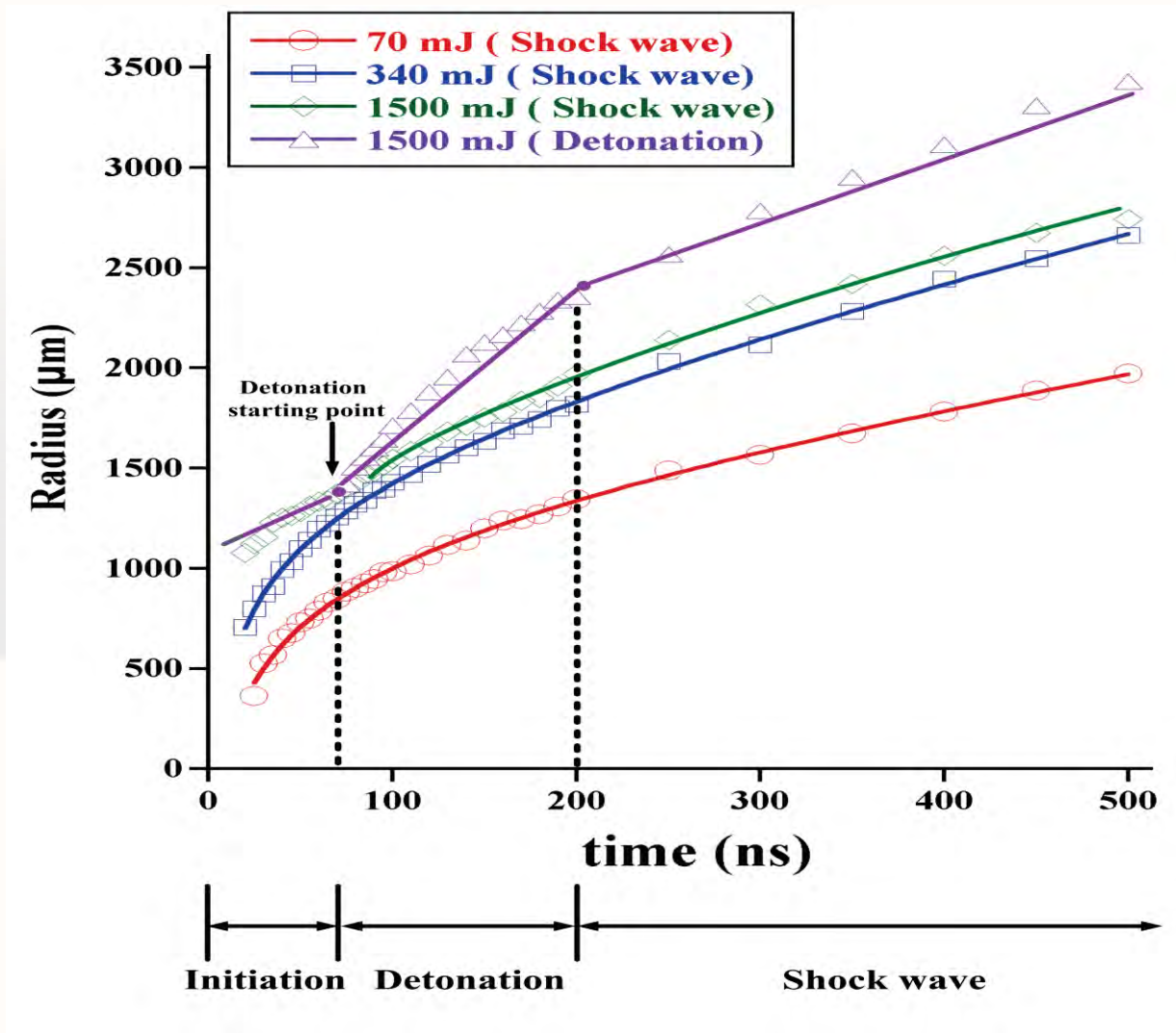
# Laser supported detonation velocity



- image resolution method (ratio of pixel and real length )

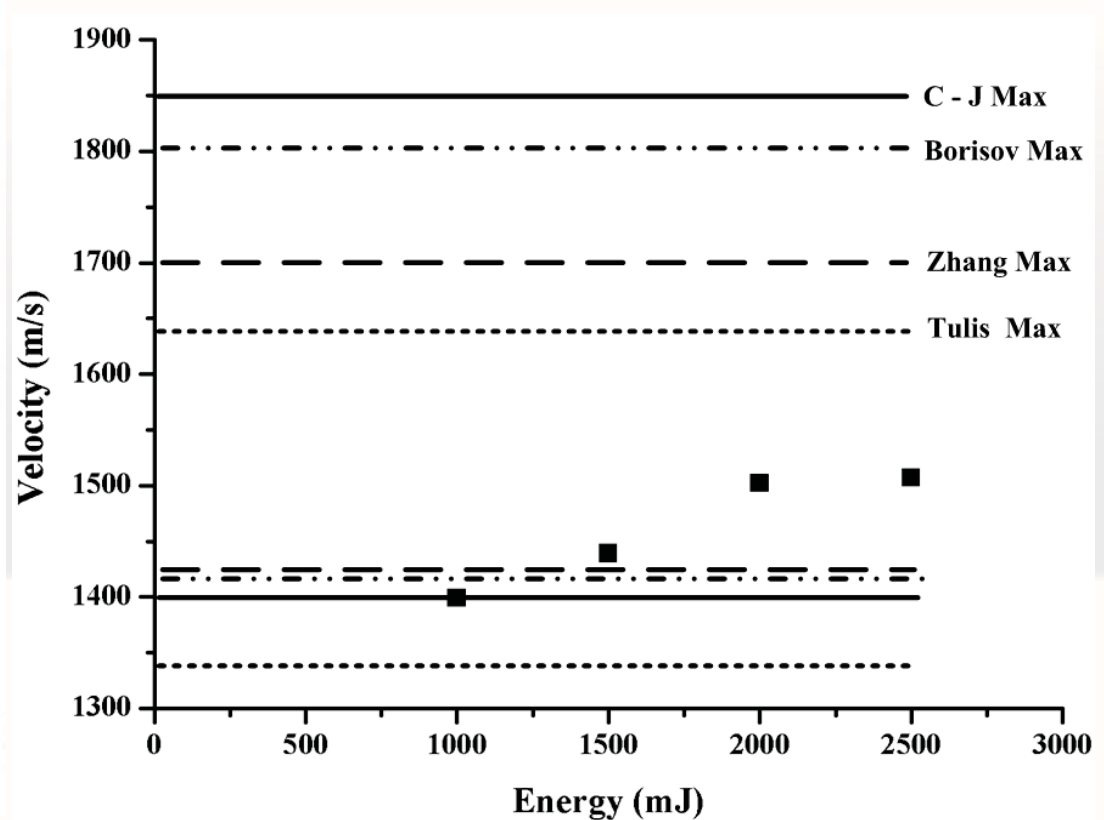
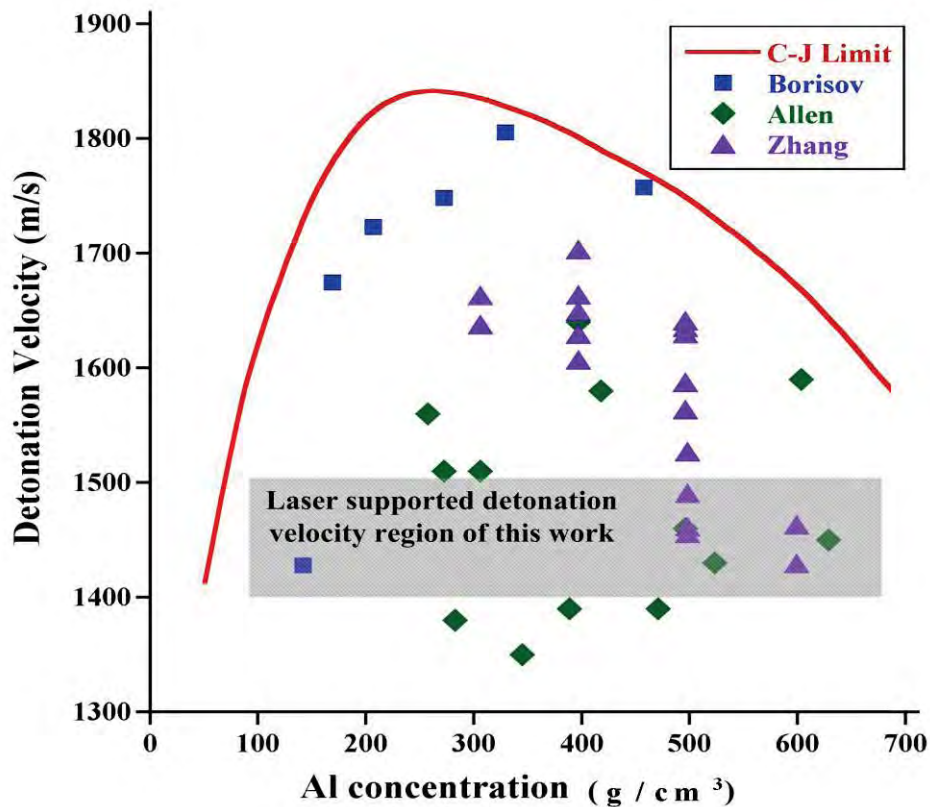
Energy (mJ)	Detonation velocity (m/s)	SD	SDOM	$\%(\frac{\sigma_{\bar{x}}}{\bar{x}} \times 100)$
1000	1399	$\pm 128$	$\pm 74$	5.27
1500	1440	$\pm 67$	$\pm 37$	5.28
2000	1502	$\pm 73$	$\pm 42$	2.79
2500	1507	$\pm 174$	$\pm 100$	6.66

- Time – Radius of propagation





- Al powder detonation vs. Laser supported detonation of Al sample comparison



# Laser supported detonation pressure

$$P_s = \frac{\rho_o V_s^2}{\gamma_{\text{metal}} + 1}$$

[ $\rho_s$ : the density of the detonation wave]

[ $P_s$ : the pressure of the detonation wave]

[ $V_s$ : the detonation wave velocity]

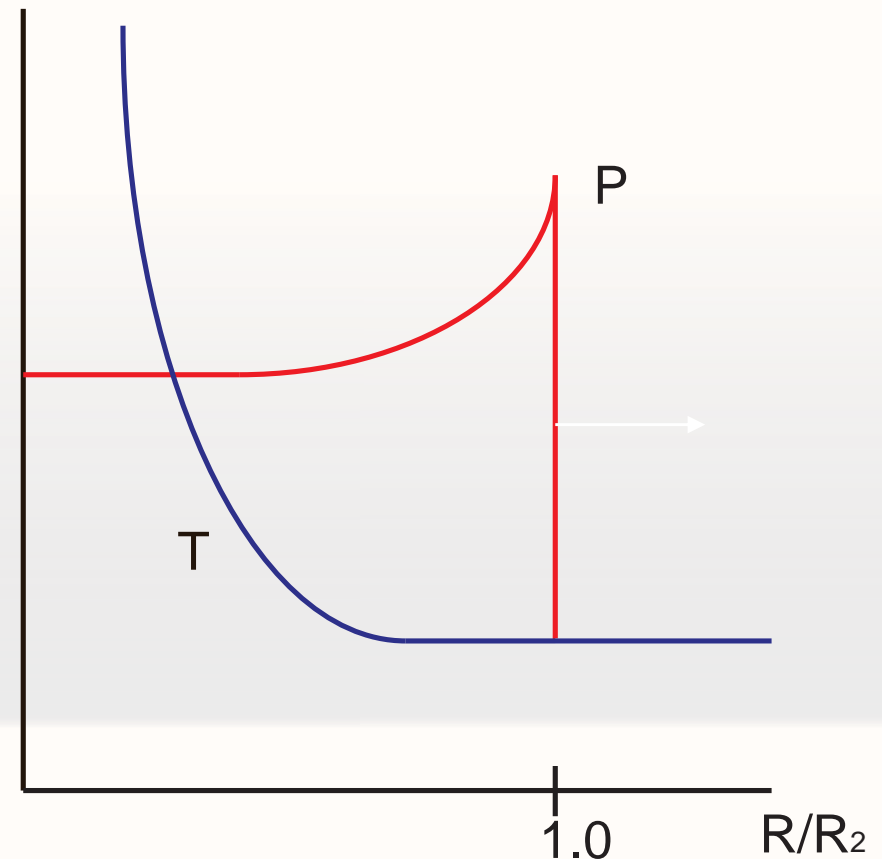
[ $\gamma_{\text{metal}}$ : the specific heat ratio of metal]

[ $\rho_o$ : the density of the ambient gas]

Energy (mJ)	Detonation pressure (pa)	SD	SDOM	%( $\frac{\sigma_{\bar{x}}}{\bar{x}} \times 100$ )
1000	981772	± 12833	± 7409	0.39
1500	1039616	± 6727	± 3884	0.71
2000	1132298	± 7281	± 4204	0.37
2500	1139751	± 17474	± 10089	0.89

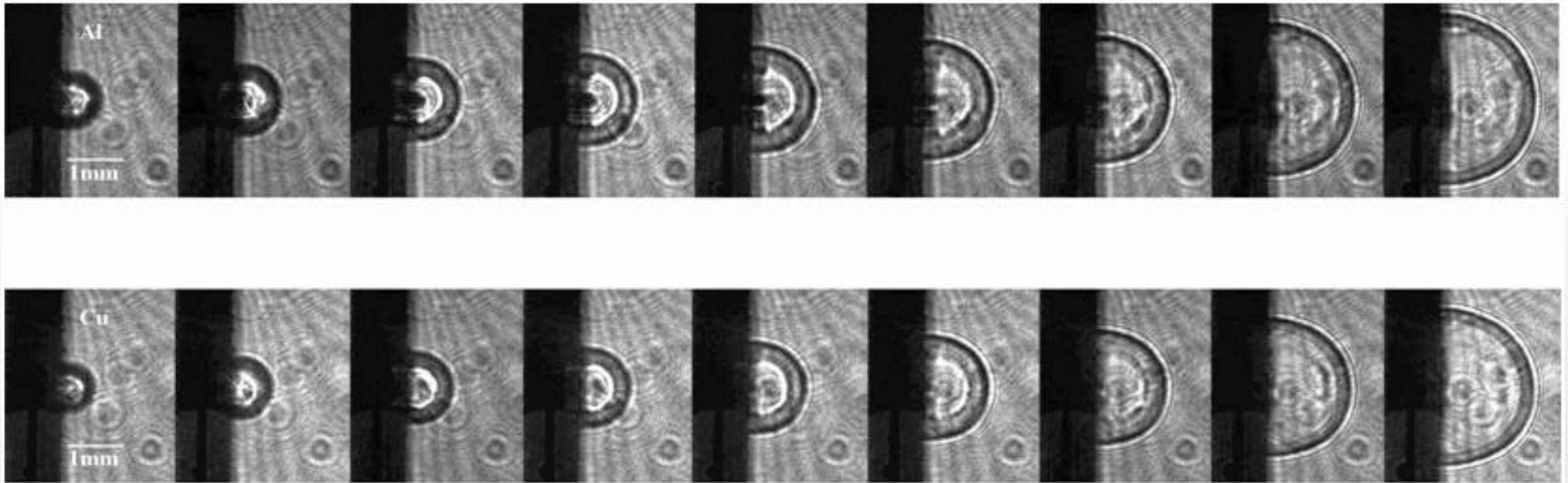
# Laser supported explosion (Sedov-Taylor Theory)

- **Explosive** melting and evaporation of solids
- Metals (with significant melt phase) undergo phase **explosion** upon high energy irradiation
- Classical Sedov-Taylor point source **explosion** theory seems to work here
- Hydrodynamic modeling of Laser **explosion**
- Ignition of energetic matter (space propulsion and weapons) via laser



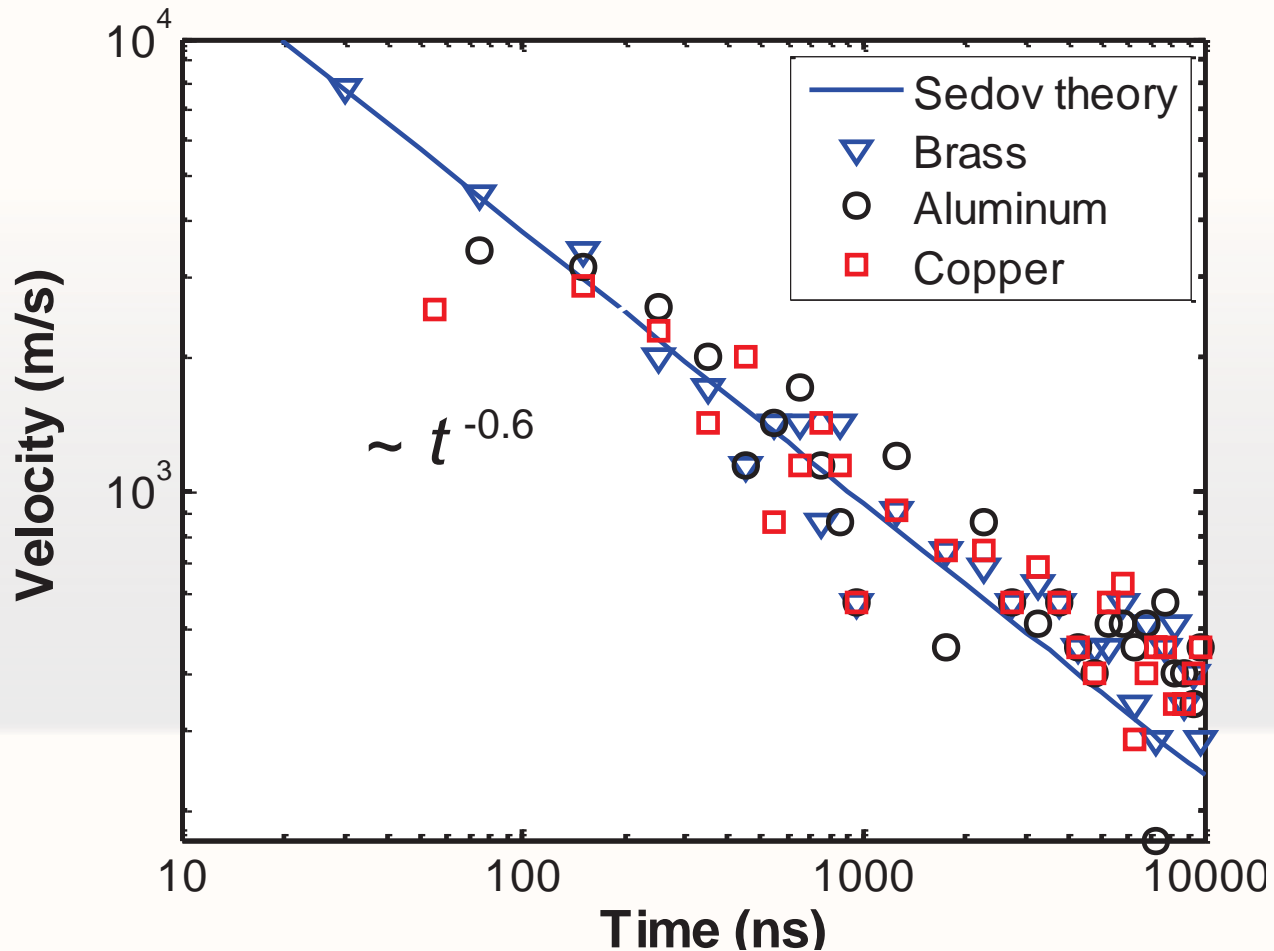
Point-source Blast Wave

# Developing spherical blast wave using laser ablation of aluminum and copper



**The Sequent shock wave images of Aluminum and Copper by 20mJ/pulse laser incident intensities shown in the increasing order of time elapse (30, 50, 70, 90, 120, 160, 200, 300, 400ns).**

# Velocity of metal phase explosion wave



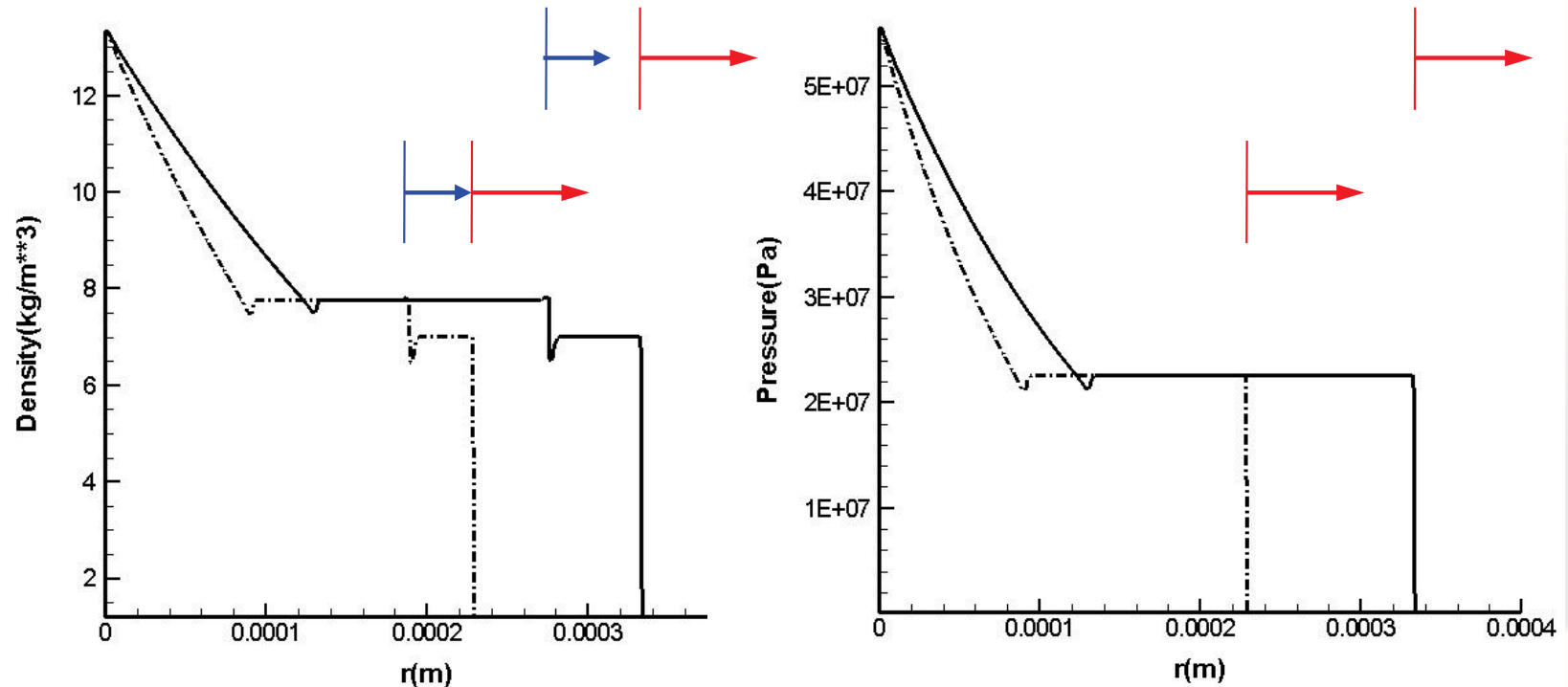
Velocities follow the theoretical  $t^{0.6}$  - curve

$$v = \lambda_0 \left( \frac{E}{\rho_{air}} \right)^{0.2} 0.4 t^{-0.6}$$

(Shown 1mm-thick metal at 200 mJ/pulse)

# Klik Micro explosion structure identified

Hot (vaporized) metal vapor / shocked (plasma) air / ambient air



- Images at two different instances (22 ns apart), suggesting shock velocity of 4500 m/s
- This says that early-on profiles are given by the solution of Euler equations for metal vapor-air shock interaction

# Structure of Riemann confirms initial blast wave profile

- **Theoretical Hugoniot properties of laser induced shock state during the first 1000 ns of 20 mJ/pulse beam.**
- **Shock speed is measured (~4500 m/s), other variables are estimated from strong shock approximation**
- **Calculations show the shocked states behind the initial planar explosive wave at 75ns is reproduced**

	<b>Shock speed</b> $u_s$ (m/s)	<b>Density<sup>1</sup></b> $\rho_s$ (kg/m <sup>3</sup> )	<b>Pressure<sup>2</sup></b> $p_s$ (MPa)
<b>Time (75 ns)</b>			
Aluminum	4571	7.76	22.5
Brass	4571		22.5
Copper	3429		12.7
<b>Time (150 ns)</b>			
Aluminum	2857	7.76	8.8
Brass	3429		12.7
Copper	3143		10.6
<b>Time (950 ns)</b>			
Aluminum	571	7.76	0.35
Brass	510		0.35
Copper	510		0.35

$$\rho_s = \rho_{air}(\gamma + 1)/(\gamma - 1), \rho_{air} = 1.29 \text{ kg/m}^3, \gamma = 1.4 \text{ for air}$$

$$p_s = 2\rho_{air}u_s^2/(\gamma + 1)$$

# Flame acceleration (DDT) criterion

$$\frac{P_2}{P_1} = \frac{2\gamma M_s^2}{\gamma + 1} - \frac{\gamma - 1}{\gamma + 1}$$

$$\frac{\rho_2}{\rho_1} = \frac{\left(\frac{\gamma + 1}{\gamma - 1}\right) \frac{P_2}{P_1} + 1}{\left(\frac{\gamma + 1}{\gamma - 1}\right) + \frac{P_2}{P_1}}$$

$$\frac{T_2}{T_1} = \frac{\left(1 + \frac{\gamma - 1}{2} M_s^2\right) \left(\frac{2\gamma}{\gamma - 1} M_s^2 - 1\right)}{\frac{(\gamma + 1)^2}{2(\gamma - 1)} M_s^2}$$

Parameter & Property	Acetylene-air mixture	Ethylene-air mixture
Initial pressure ( $P_0$ )	$1.33 \times 10^4 \text{Pa}$	$1.33 \times 10^4 \text{Pa}$
Initial temperature ( $T_0$ )	293K	293K
Initial density ( $\rho_0$ )	$1.58 \times 10^{-1} \text{kg/m}^3$	$1.58 \times 10^{-1} \text{kg/m}^3$
Specific heat rate ( $\gamma$ )	1.25	1.15
Molecular weight (M)	$29 \times 10^{-3} \text{kg/mol}$	$29 \times 10^{-3} \text{kg/mol}$
Pre-exponential factor (A)	$1 \times 10^9 \text{m}^3/(\text{kg s})$	$3.2 \times 10^8 \text{m}^3/(\text{kg s})$
Activation energy ( $E_a$ )	$29.3RT_0$	$35.3514RT_0$
Chemical energy release (q)	$35.0RT_0/M$	$48.824RT_0/M$
Transport constants ( $\nu_0 = \mu_0 = D_0$ )	$1.3 \times 10^{-7} \text{kg}/(\text{s m K}^{0.7})$	$7.0 \times 10^{-8} \text{kg}/(\text{s m K}^{0.7})$
Adiabatic flame temperature	2340K	2625K
Adiabatic flame density	$1.98 \times 10^{-2} \text{kg/m}^3$	$1.77 \times 10^{-2} \text{kg/m}^3$
Incident shock intensity	1.2	1.9

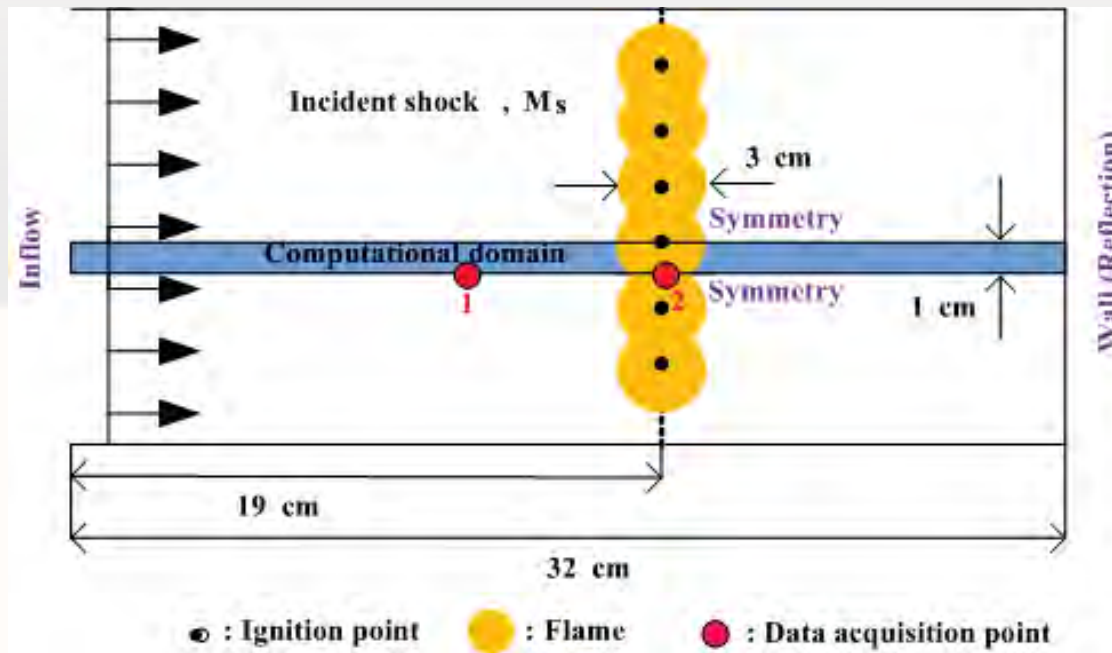


# Acetylene-air mixture



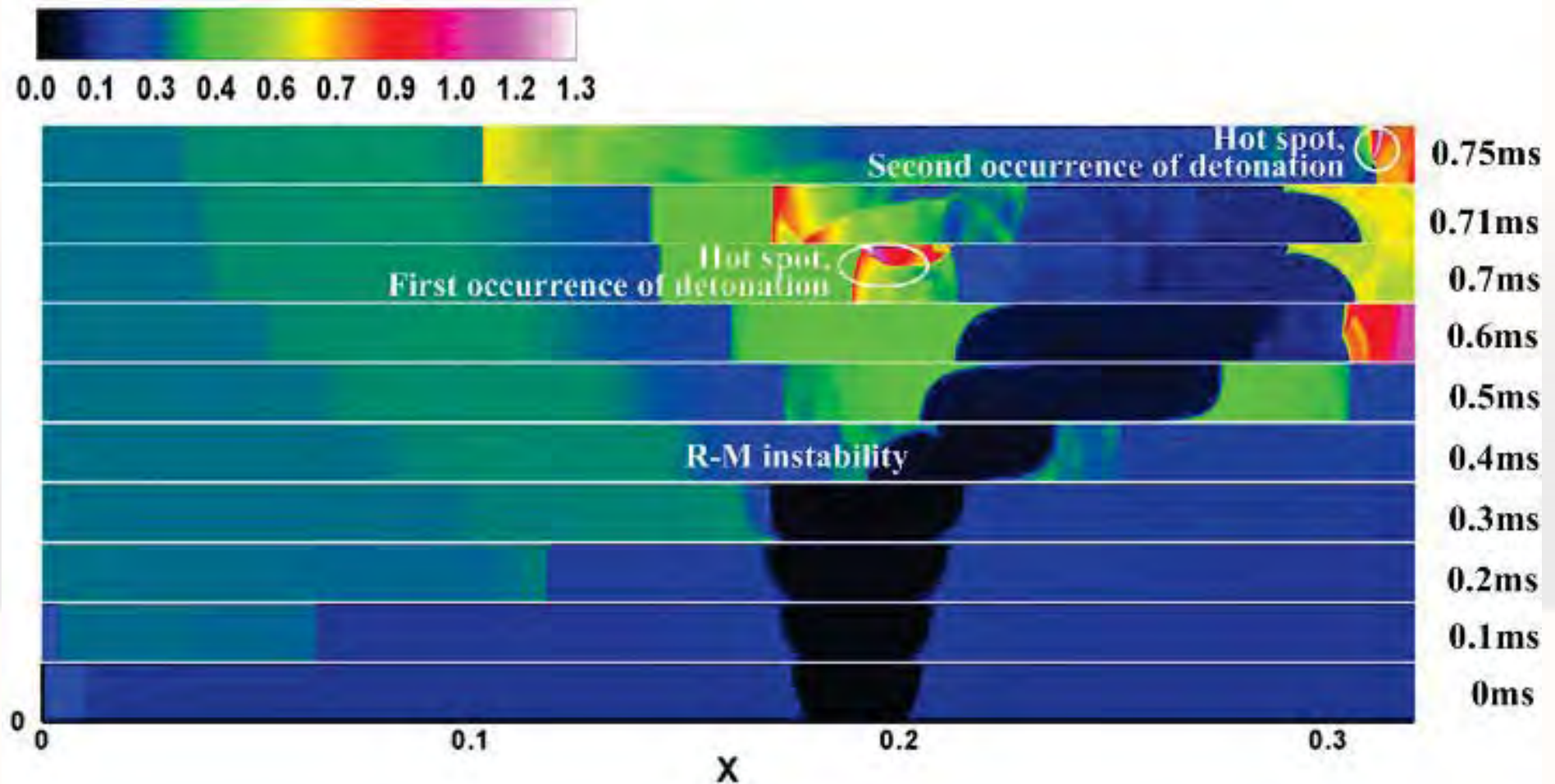
Experimental schlieren image

[T. Scarinci, etc., 1993]



2D numerical domain

# Acetylene-air mixture

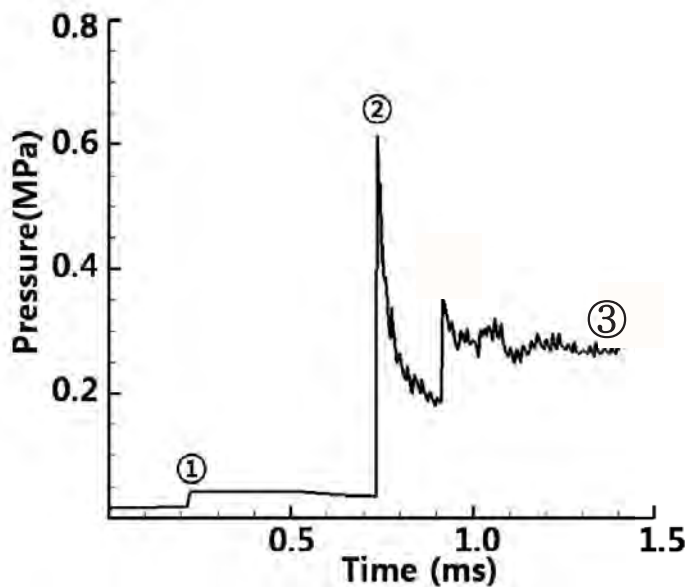


Time sequences of density fields under  $M=1.2$  incident shock interacting with an acetylene-air flame

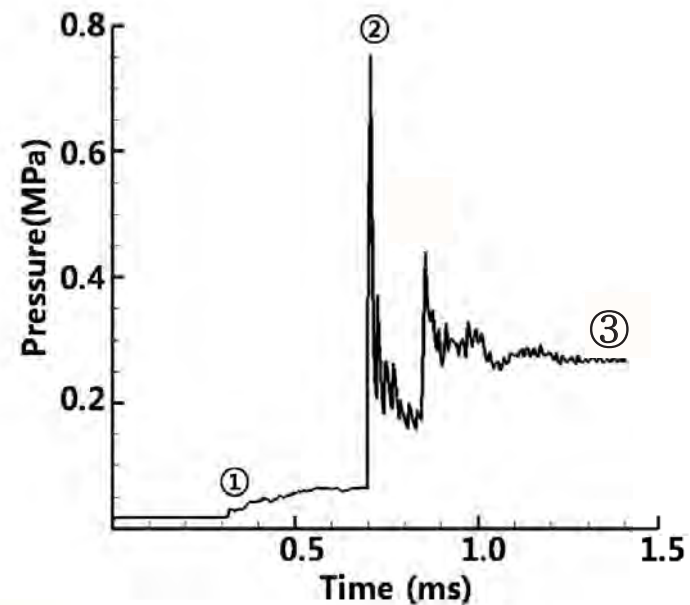
# Calculated pressure histories

- Gages

- ① Incident shock wave propagates from left to right.
- ② Pressure peak shows detonation occurrence.  
(peak pressure: 0.6~0.8MPa, velocity: 1887m/s)
- ③ Pressure is stabilized after detonation (0.27MPa).

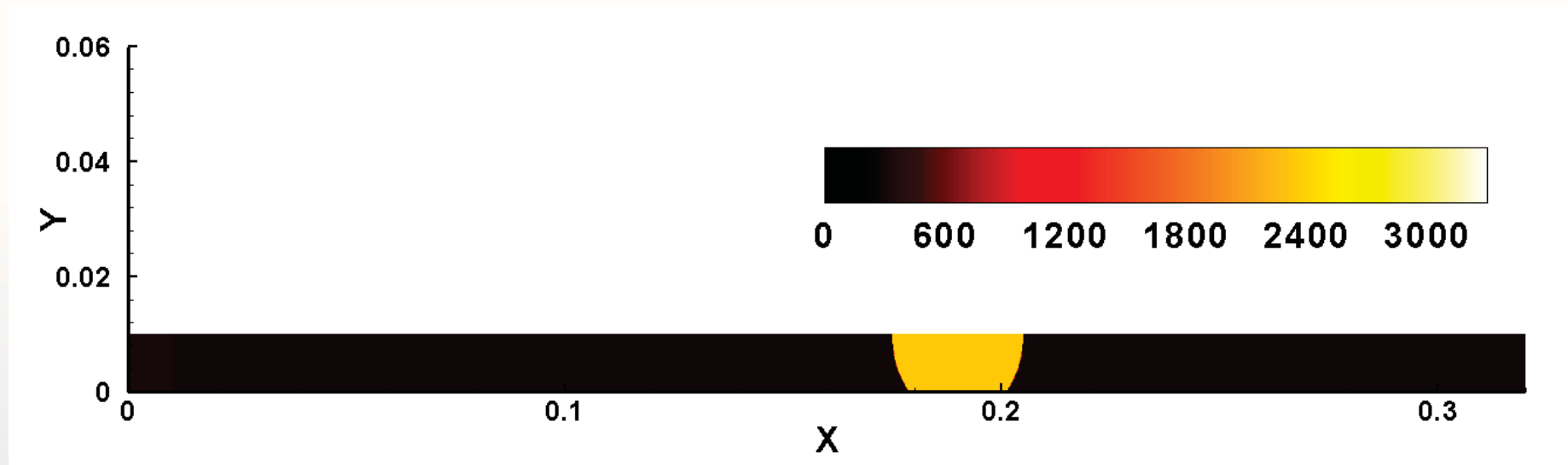


Pressure history at gage 1



Pressure history at gage 2

# Acetylene-air mixture



- Temperature

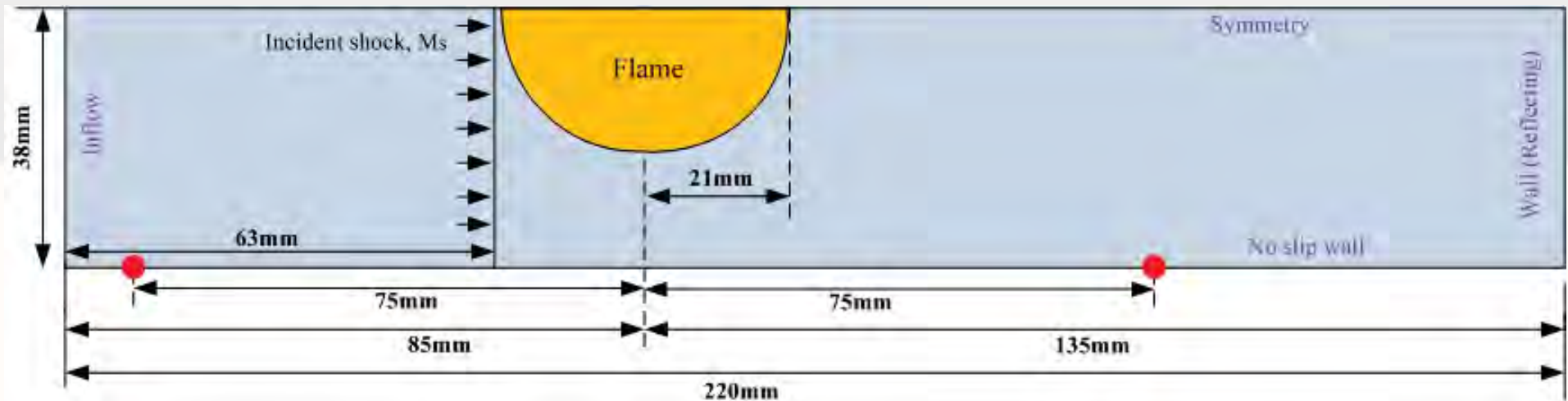
	Reference value	Numerical value
Peak pressure	805509Pa [0.8MPa] [ <i>T. Scarinci, 1993</i> ]	0.6~0.8MPa
Pressure ratio	19.5 [ <i>I. Glassman, 1996</i> ]	20.3
Detonation velocity	CJ Detonation velocity, 1870m/s [ <i>A. M. Khokhlov, 1999</i> ]	1887m/s

# Ethylene-air mixture – Initial Setup



Experimental schlieren image

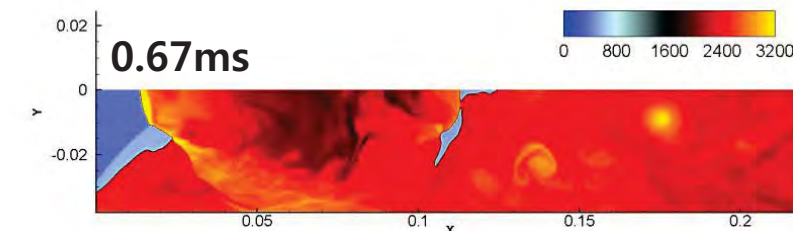
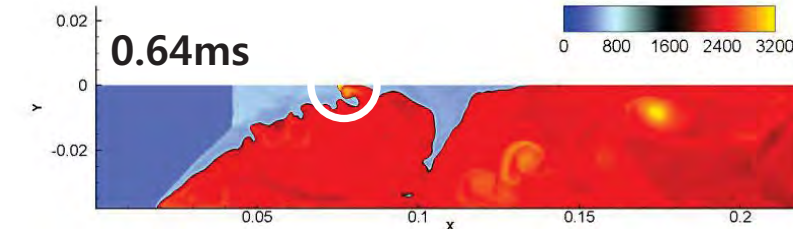
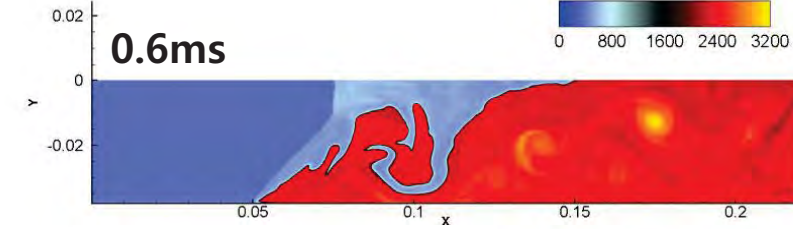
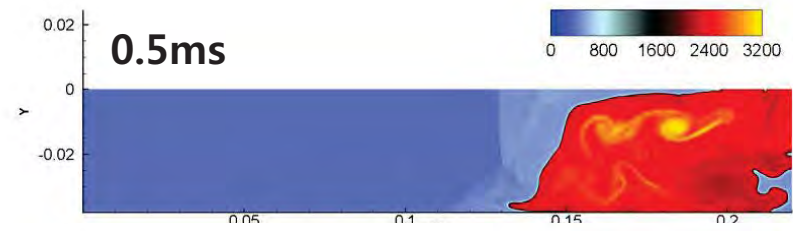
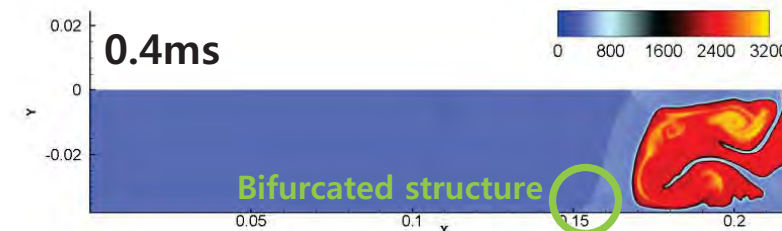
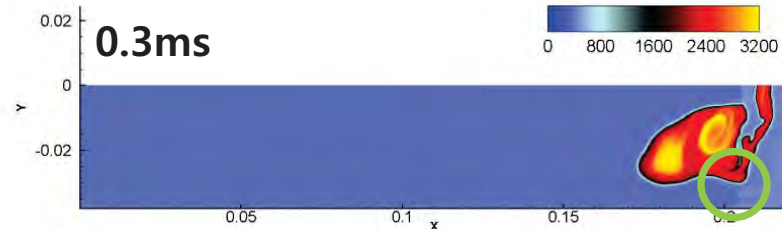
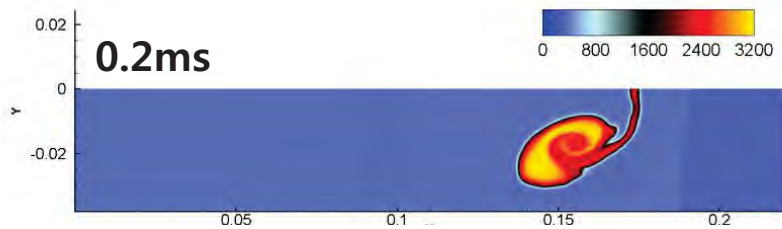
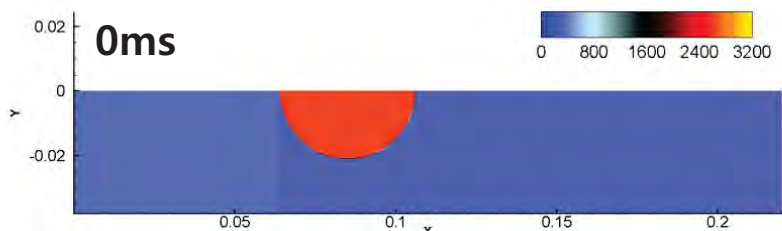
[G. Thomas, etc., 2001]



2D numerical domain

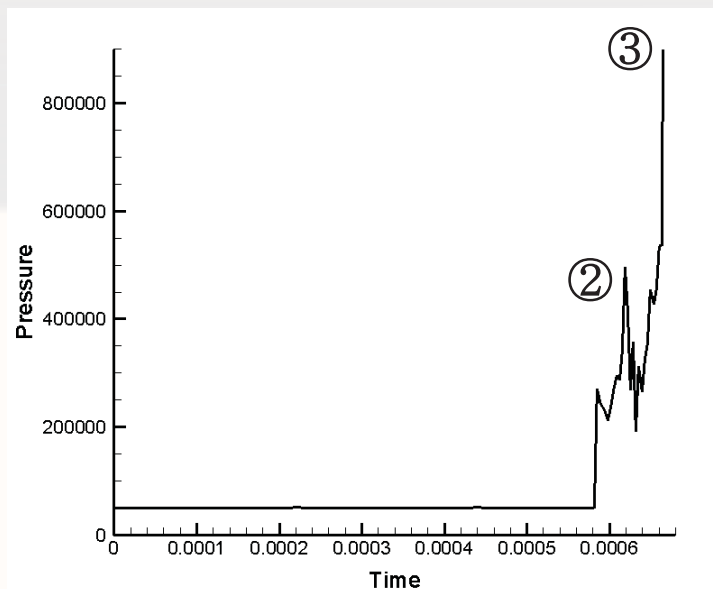
# Temperature: O: detonation occurrence)

- Flame surface is increased by flame instability (0.4ms)
- Bifurcated structure is formed by no slip boundary condition.
- Detonation is occurred at 0.64ms.

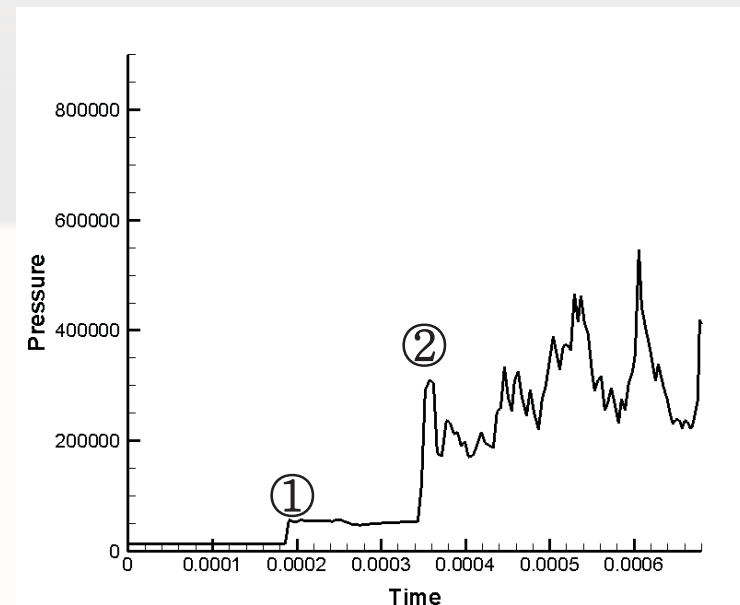


# Pressure histories

- ① Incident shock wave(0.05MPa, 0.4825kg/m<sup>3</sup>, 363K) propagates from left to right.
- ② Pressure is increased by propagation of strange wave.  
(Strange wave: a kind of reflection wave that is occurred between flame and wall and accompanies turbulent flame)  
velocity~1500m/s, pressure ratio~10 [*G. Thomas, et al., 2001*]
- ③ Pressure peak(more than 0.8MPa) shows detonation occurrence.

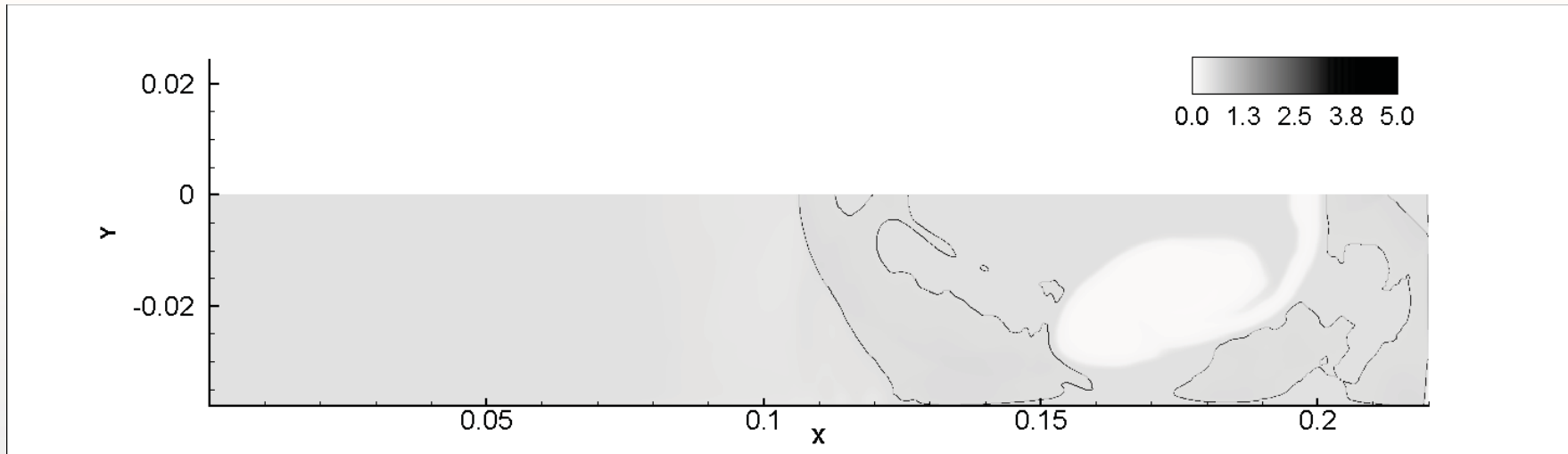


Pressure history at gage 1

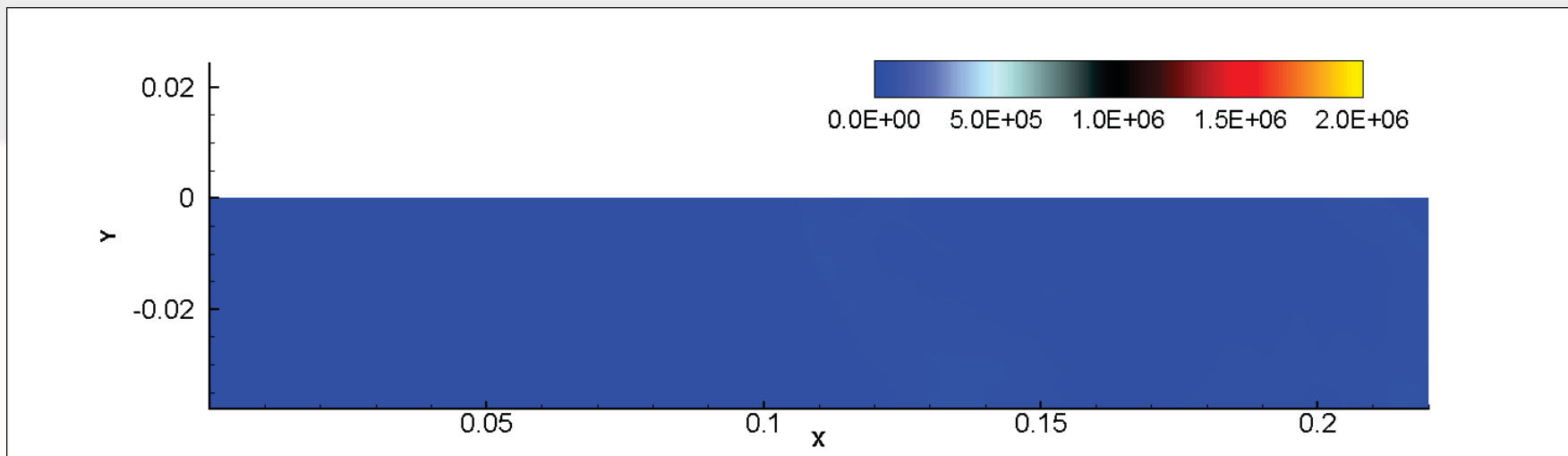


Pressure history at gage 2

# Density and Pressure (0.2~0.7 ms)



**Video clip of density field [unit: kg/m<sup>3</sup>]**



**Video clip of pressure field [unit: Pa]**



- **Hydrocode** development for simulation of energetic materials
  - Cook off test of Rocket Propellants, Explosives, Gaseous Fuels
  - Many more simulation capabilities not shown
- High power laser ablation experiment for studying **metal detonation in air**: Not in powder form, rather in a bulk.
- Flame acceleration research via DNS for understanding the origin of **Deflagration-to-Detonation** transition from a quiescent flame: rocket chamber instability, engine knock, gas pipe failure due to explosions, etc.